

On the equivalence between the energy and virial routes to the equation of state of hard-sphere fluids

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The energy route to the equation of state of hard-sphere fluids is ill defined since the internal energy is just that of an ideal gas, and thus it is independent of density. It is shown that this ambiguity can be avoided by considering a square-shoulder interaction and taking the limit of vanishing shoulder width. The resulting hard-sphere equation of state coincides exactly with the one obtained through the virial route. Therefore, the energy and virial routes to the equation of state of hard-sphere fluids can be considered as equivalent. © 2005 American Institute of Physics. [DOI: 10.1063/1.1992469]

Given a fluid of particles interacting via a two-body potential $\varphi(r)$, its equation of state (EOS) can be obtained in terms of the radial distribution function $g(r; \rho, \beta)$, where ρ is the number density and $\beta = 1/k_B T$ is the inverse temperature, through a number of routes.¹ The most common ones are the virial route,

$$\frac{\beta p}{\rho} \equiv Z(\rho, \beta) = 1 + 2^{d-1} v_d \rho \int_0^\infty dr r^d y(r; \rho, \beta) \frac{\partial}{\partial r} e^{-\beta \varphi(r)}, \quad (1)$$

the compressibility route

$$\left(\beta \frac{\partial p}{\partial \rho} \right)^{-1} \equiv \chi(\rho, \beta) = 1 + 2^d d v_d \rho \times \int_0^\infty dr r^{d-1} [h(r; \rho, \beta) - 1], \quad (2)$$

and the energy route

$$u(\rho, \beta) = \frac{d}{2\beta} \left[1 + 2^d v_d \rho \beta \int_0^\infty dr r^{d-1} \varphi(r) e^{-\beta \varphi(r)} y(r; \rho, \beta) \right]. \quad (3)$$

In Eqs. (1)–(3), p is the pressure, Z is the compressibility factor, d is the dimensionality of the system, $v_d = (\pi/4)^{d/2} / \Gamma(1+d/2)$ is the volume of a d -dimensional sphere of unit diameter, χ is the isothermal susceptibility, u is the internal energy per particle, $y(r) \equiv \exp[\beta \varphi(r)] g(r)$ is the cavity function, and $h(r) \equiv g(r) - 1$ is the total correlation function. If the *exact* function $g(r; \rho, \beta)$ is inserted, the three routes are thermodynamically consistent, i.e.,

$$\chi^{-1}(\rho, \beta) = \frac{\partial}{\partial \rho} [\rho Z(\rho, \beta)], \quad (4)$$

$$\rho \frac{\partial}{\partial \rho} u(\rho, \beta) = \frac{\partial}{\partial \beta} Z(\rho, \beta). \quad (5)$$

On the other hand, if an *approximate* function $g(r; \rho, \beta)$ is used, the compressibility factor obtained directly from Eq. (1) does not necessarily coincide with that obtained from the combination of Eqs. (2) and (4) or with that obtained from the combination of Eqs. (3) and (5).

Let us now particularize to the hard-sphere (HS) interaction

$$\varphi_{\text{HS}}(r) = \begin{cases} \infty, & r < \sigma, \\ 0, & r > \sigma. \end{cases} \quad (6)$$

In that case, $y(r; \rho, \beta) = y_{\text{HS}}(r; \rho \sigma^d)$ and Eq. (1) becomes

$$Z_{\text{HS}}(\rho \sigma^d) = 1 + 2^{d-1} v_d \rho \sigma^d y_{\text{HS}}(\sigma; \rho \sigma^d). \quad (7)$$

On the other hand, Eq. (3) is reduced to

$$u_{\text{HS}}(\beta) = \frac{d}{2\beta}. \quad (8)$$

Since Z_{HS} is independent of temperature and u_{HS} is independent of density, Eq. (5) is trivially satisfied as $0=0$, and so it is not possible to get the compressibility factor from the internal energy.

A way of circumventing the ill definition of the energy route to the EOS of an HS fluid consists in considering a convenient interaction potential that encompasses that of hard spheres in certain limits but for which the temperature plays a relevant role. The simplest choice for such a potential is perhaps the so-called square-shoulder (SS) potential:²

$$\varphi_{\text{SS}}(r) = \begin{cases} \infty, & r < \sigma, \\ \epsilon, & \sigma < r < \sigma', \\ 0, & r > \sigma', \end{cases} \quad (9)$$

where ϵ is a positive constant. For this potential, Eq. (3) becomes

$$u_{\text{SS}}(\rho, \beta) = \frac{d}{2\beta} \left[1 + 2^d v_d \rho \beta \epsilon e^{-\beta \epsilon} \int_\sigma^{\sigma'} dr r^{d-1} y_{\text{SS}}(r; \rho, \beta) \right]. \quad (10)$$

The SS interaction is equivalent to an HS interaction of diameter σ in the infinite-temperature limit ($\beta \epsilon \rightarrow 0$) and to an

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HS interaction of diameter σ' in the zero-temperature limit ($\beta\epsilon \rightarrow \infty$). Of course, the equivalence $SS \rightarrow HS$ also holds in the limit of the zero shoulder width ($\sigma' \rightarrow \sigma$).

Let us now imagine that $y_{SS}(r; \rho, \beta)$ is known, either exactly or approximately (e.g., as obtained from the Percus-Yevick approximation), and thus it is possible to compute $u_{SS}(\rho, \beta)$ from Eq. (10). Then, we can get $Z_{SS}(\rho, \beta)$ from Eq. (5) as

$$\begin{aligned} Z_{SS}(\rho, \beta) - Z_{HS}(\rho\sigma^d) \\ = 2^{d-1} d v_d \epsilon \rho \frac{\partial}{\partial \rho} \int_0^\beta d\beta' e^{-\beta'\epsilon} \int_\sigma^{\sigma'} dr r^{d-1} y_{SS}(r; \rho, \beta'), \end{aligned} \quad (11)$$

where we have made use of the property $\lim_{\beta\epsilon \rightarrow 0} Z_{SS}(\rho, \beta) = Z_{HS}(\rho\sigma^d)$. Taking into account the complementary condition $\lim_{\beta\epsilon \rightarrow \infty} Z_{SS}(\rho, \beta) = Z_{HS}(\rho\sigma'^d)$, Eq. (11) yields

$$\begin{aligned} \frac{Z_{HS}(\rho\sigma'^d) - Z_{HS}(\rho\sigma^d)}{\rho\sigma'^d - \rho\sigma^d} \\ = \frac{2^{d-1} d v_d \epsilon}{\sigma'^d - \sigma^d} \frac{\partial}{\partial \rho} \int_0^\infty d\beta e^{-\beta\epsilon} \int_\sigma^{\sigma'} dr r^{d-1} y_{SS}(r; \rho, \beta). \end{aligned} \quad (12)$$

Equation (12) can be considered as the condition defining the compressibility factor of an HS fluid associated with the energy route. To proceed further, we take the limit $\sigma' \rightarrow \sigma$. Thus,

$$\lim_{\sigma' \rightarrow \sigma} \frac{Z_{HS}(\rho\sigma'^d) - Z_{HS}(\rho\sigma^d)}{\rho\sigma'^d - \rho\sigma^d} = \sigma^{-d} \frac{\partial}{\partial \rho} Z_{HS}(\rho\sigma^d), \quad (13)$$

$$\lim_{\sigma' \rightarrow \sigma} \frac{1}{\sigma'^d - \sigma^d} \int_\sigma^{\sigma'} dr r^{d-1} y_{SS}(r; \rho, \beta) = \frac{1}{d} y_{HS}(\sigma; \rho\sigma^d). \quad (14)$$

Therefore, Eq. (12) is reduced in the limit $\sigma' \rightarrow \sigma$ to

$$\frac{\partial}{\partial \rho} Z_{HS}(\rho\sigma^d) = 2^{d-1} d v_d \frac{\partial}{\partial \rho} \rho\sigma^d y_{HS}(\sigma; \rho\sigma^d). \quad (15)$$

Finally, integrating over the density and imposing the ideal gas boundary condition $Z_{HS}(0)=1$, the virial EOS (7) is re-obtained.

The generalization to mixtures (either additive or nonadditive) is straightforward. In that case, the SS potential for

each pair ij is defined by Eq. (9) with the changes $\epsilon \rightarrow \epsilon_{ij}$, $\sigma \rightarrow \sigma_{ij}$, and $\sigma' \rightarrow \sigma'_{ij} = \lambda \sigma_{ij}$, where the factor λ is common to all the pairs. The equation equivalent to Eq. (12) in the case of mixtures is

$$\begin{aligned} \frac{Z_{HS}(\rho\sigma_{\text{eff}}'^d) - Z_{HS}(\rho\sigma_{\text{eff}}^d)}{\rho(\lambda^d - 1)} \\ = \frac{2^{d-1} d v_d}{\lambda^d - 1} \frac{\partial}{\partial \rho} \rho \sum_{ij} x_i x_j \epsilon_{ij} \int_0^\infty d\beta e^{-\beta\epsilon_{ij}} \\ \times \int_{\sigma_{ij}}^{\sigma'_{ij}} dr r^{d-1} y_{ij}^{SS}(r; \rho, \beta), \end{aligned} \quad (16)$$

where $\{x_i\}$ are mole fractions and $\sigma_{\text{eff}}^d \equiv \sum_{i,j} x_i x_j \sigma_{ij}^d$, $\sigma_{\text{eff}}'^d \equiv \sum_{i,j} x_i x_j \sigma_{ij}'^d = \lambda^d \sigma_{\text{eff}}^d$. Taking the limit $\lambda \rightarrow 1$ in Eq. (16), one gets

$$Z_{HS}(\rho\sigma_{\text{eff}}^d) = 1 + 2^{d-1} d v_d \rho \sum_{ij} x_i x_j \sigma_{ij}^d y_{ij}^{HS}(\sigma_{ij}; \rho\sigma_{\text{eff}}^d), \quad (17)$$

which is not but the virial EOS for an HS mixture.

In summary, in this note I have shown that the ill definition of the energy route to the EOS of an HS fluid is avoided by first considering an SS fluid and then taking the limit of the vanishing shoulder width. The resulting EOS coincides exactly with the one obtained through the virial route. From that point of view, the energy and virial routes to the EOS of HS fluids can be considered as equivalent.

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