Can a system of *elastic* hard spheres mimic the transport properties of a granular gas?

A. Astillero and <u>A. Santos</u> Universidad de Extremadura, Badajoz, Spain

Direct Simulation Monte Carlo: The Past 40 Years and the Future Politecnico di Milano June 3, 2003

OUTLINE

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- II. "Equivalent" system of *elastic* hard spheres
- **III.** Simple shear flow
- IV. Applications and extensions
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I. Inelastic hard spheres

• The prototype model for a granular medium in the rapid flow regime is a gas of *inelastic* hard spheres with a constant coefficient of normal restitution α .



FIG. 1: Sketch of inelastic collisions (after T.P.C. van Noije & M.H. Ernst).

• Direct collision:

$$\widehat{b}\mathbf{v}_{1} \equiv \mathbf{v}_{1}^{*} = \mathbf{v}_{1} - \frac{1+\alpha}{2} \left(\mathbf{v}_{12} \cdot \widehat{\boldsymbol{\sigma}}\right) \widehat{\boldsymbol{\sigma}} \\ \widehat{b}\mathbf{v}_{2} \equiv \mathbf{v}_{2}^{*} = \mathbf{v}_{2} + \frac{1+\alpha}{2} \left(\mathbf{v}_{12} \cdot \widehat{\boldsymbol{\sigma}}\right) \widehat{\boldsymbol{\sigma}}$$

• Restituting collision:

$$\widehat{b}^{-1}\mathbf{v}_1 \equiv \mathbf{v}_1^{**} = \mathbf{v}_1 - \frac{1+\alpha^{-1}}{2} \left(\mathbf{v}_{12} \cdot \widehat{\boldsymbol{\sigma}} \right) \widehat{\boldsymbol{\sigma}} \\ \widehat{b}^{-1}\mathbf{v}_2 \equiv \mathbf{v}_2^{**} = \mathbf{v}_2 + \frac{1+\alpha^{-1}}{2} \left(\mathbf{v}_{12} \cdot \widehat{\boldsymbol{\sigma}} \right) \widehat{\boldsymbol{\sigma}}$$

Boltzmann equation (molecular chaos)

$$(\partial_t + \mathbf{v}_1 \cdot \nabla) f(\mathbf{v}_1) = J^{(\alpha)}[\mathbf{v}_1|f],$$

$$J^{(\alpha)}[\mathbf{v}_1|f] = \sigma^{d-1} \int \mathrm{d}\mathbf{v}_2 \int \mathrm{d}\widehat{\boldsymbol{\sigma}} \,\Theta(\mathbf{v}_{12} \cdot \widehat{\boldsymbol{\sigma}})(\mathbf{v}_{12} \cdot \widehat{\boldsymbol{\sigma}}) \left(\alpha^{-2}\widehat{b}^{-1} - 1\right) f(\mathbf{v}_1) f(\mathbf{v}_2).$$

• Conservation of mass:

$$m \int \mathrm{d}\mathbf{v} \, J^{(\alpha)}[\mathbf{v}|f] = 0.$$

• Conservation of momentum:

$$m\int \mathrm{d}\mathbf{v}\,\mathbf{v}J^{(\alpha)}[\mathbf{v}|f]=\mathbf{0}.$$

• Energy decrease (collisional "cooling"):

$$\frac{m}{dn} \int d\mathbf{v} V^2 J^{(\alpha)}[\mathbf{v}|f] = -\zeta(\alpha)T, \quad \mathbf{V} \equiv \mathbf{v} - \mathbf{u},$$

$$T = \frac{m}{d} \langle V^2 \rangle$$
: Granular temperature,

$$\zeta(\alpha) \propto (1 - \alpha^2) \langle V_{12}^3 \rangle$$
: Cooling rate.

• Local equilibrium approximation:

$$\zeta(\alpha) \to \zeta_0(\alpha) = \frac{d+2}{4d}\nu_0(1-\alpha^2),$$

 $\nu_0 \propto n \sigma^{d-1} (2T/m)^{1/2}$: (effective) collision frequency.

II. "Equivalent" system of *elastic* hard spheres

• *Inelastic* hard spheres (IHS):

$$\left. \frac{\partial T}{\partial t} \right|_{\text{coll}} = -\zeta(\alpha)T.$$

• *Elastic* hard spheres (EHS):

$$\left. \frac{\partial T}{\partial t} \right|_{\text{coll}} = 0$$

But . . .

if a drag or friction force $\mathbf{F}_{\text{drag}} = -m\gamma \mathbf{V}$ exists, then

$$\left. \frac{\partial T}{\partial t} \right|_{\text{friction}} = -2\gamma T.$$

• Can a system of EHS with $\gamma = \frac{1}{2}\zeta_0(\alpha) \simeq \frac{1}{2}\zeta(\alpha)$ mimic the properties of a system of IHS?

$$\underbrace{J^{(\alpha)}[\mathbf{v}|f]}_{\text{inelastic collisions}} \to \beta(\alpha) \underbrace{J^{(1)}[\mathbf{v}|f]}_{\text{elastic collisions}} + \underbrace{\frac{1}{2}\zeta_0(\alpha)\frac{\partial}{\partial\mathbf{v}}\cdot(\mathbf{V}f)}_{\text{friction}}$$

 $\beta(\alpha)$: parameter to modify the collision frequency of EHS relative to that of IHS ($\sigma_{\text{EHS}} \neq \sigma_{\text{IHS}}$).

• Both systems (IHS and EHS+friction) yield the same hydrodynamic balance equations (except for the approximation $\zeta \to \zeta_0$).

However, the *microscopic* dynamics is quite different.



FIG. 2: Sketch of IHS and EHS

Homogeneous cooling state

• Haff's law:

$$T = \frac{T_0}{\left(1 + \frac{1}{2}\zeta_{t=0}t\right)^2}.$$

• Scaling solution:

$$f^*(\mathbf{c}) = \frac{1}{n} \left(\frac{2T}{m}\right)^{d/2} f(\mathbf{v}), \quad \mathbf{c} = \mathbf{v}/(2T/m)^{1/2}.$$

• EHS:

$$f^*(\mathbf{c}) = \pi^{-d/2} e^{-c^2}$$

• IHS:

$$f^*(\mathbf{c}) \neq \pi^{-d/2} e^{-c^2},$$

second cumulant (kurtosis): $a_2 \equiv \frac{4}{d(d+2)} \langle c^4 \rangle - 1 \neq 0$,

high energy tail: $f^*(\mathbf{c}) \sim e^{-Ac} \Rightarrow G(\mathbf{c}) \equiv e^{Ac} f^*(\mathbf{c}) \to \text{const.}$



FIG. 3: d=3 [J. M. Montanero & A.S., Gran. Matt. $\mathbf{2},\,53~(2000)]$

• What happens in *inhomogeneous* states?

Transport coefficients (Chapman–Enskog method)

- So far, $\beta(\alpha)$ remains undetermined.
- A comparison between the transport coefficients for IHS and EHS shows that the optimal choice for the *shear viscosity* is

$$\beta(\alpha) = \frac{1+\alpha}{2} \left[1 - \frac{d-1}{2d} (1-\alpha) \right] \equiv \beta_{\eta}(\alpha),$$

while the optimal choice for the *thermal conductivity* is

$$\beta(\alpha) = \frac{1+\alpha}{2} \left[1 + \frac{3}{8} \frac{4-d}{d-1} (1-\alpha) \right] \equiv \beta_{\kappa}(\alpha).$$

• This suggests to take

$$\beta(\alpha) = \frac{1+\alpha}{2}$$

as the *simplest* choice.



FIG. 4: Shear viscosity and thermal conductivity (d = 3).

• Steady state:



FIG. 5: Sketch of the simple shear flow

$$a = \frac{U}{L} = (\text{constant}) \text{ shear rate.}$$

• Energy balance equation:

$$\frac{\partial}{\partial t}T = \underbrace{-\frac{2}{d}aP_{xy}}_{\text{viscous heating}} + \underbrace{(-\zeta T)}_{\text{inelastic cooling (or friction)}} = 0,$$

 $P_{ij} = mn \langle V_i V_j \rangle$: pressure tensor.

• Test case:

d = 3,

 $a = 4\tau_0^{-1}, \quad \tau_0 = \text{initial m.f.t. of the IHS gas},$

 $L = 2.5\lambda$, $\lambda = average m.f.p.$ of the IHS gas,

$$U = aL = 10v_0, \quad v_0 = \lambda/\tau_0 =$$
initial thermal velocity

- Initial conditions:
 - Total equilibrium.
 - Local equilibrium.

• DSMC details:

$$N = 10^4$$
 particles,

50 layers
$$\Rightarrow \Delta L = 0.05\lambda$$

Time step:
$$\delta t = 10^{-3} \tau_0 \sqrt{T_0/T}$$

- IHS: Inelastic collisions with $\alpha = 0.9$.
- EHS: Elastic collisions (rate reduced by a factor $\beta = \frac{1+\alpha}{2} = 0.95$) + Friction.



FIG. 6: Number of collisions per particle

• Initial condition: Total equilibrium



FIG. 7: Velocity profiles.



FIG. 8: Temperature profiles.



FIG. 9: Density profiles.

• Initial condition: Local equilibrium



FIG. 10: Time evolution of the temperature and the pressure tensor.



FIG. 11: Time evolution of $\langle V_{12}^3 \rangle / \langle V_{12}^3 \rangle_0$, and the second and third cumulants.

$$a_2 = \frac{4}{15} \langle C^4 \rangle - 1, \quad -a_3 = \frac{8}{105} \langle C^4 \rangle - 1 - 3a_2.$$



FIG. 12: Steady-state distribution function for the magnitude of $\mathbf{C} = \mathbf{V}/\sqrt{2T/m}$.

$$F(C) = C^2 \int d\widehat{\mathbf{C}} f^*(\mathbf{C}), \quad F_0(C) = 4\pi^{-1/2}C^2e^{-C^2}.$$

IV. Applications and extensions Kinetic modeling

- The mapping IHS↔EHS allows one to extend to granular gases those kinetic models originally proposed for conventional gases.
- Bhatnagar–Gross–Krook (BGK) model:

$$J^{(1)}[\mathbf{v}|f] \to -\nu_0[f(\mathbf{v}) - f_0(\mathbf{v})], \quad f_0(\mathbf{v}) = n \left(\frac{m}{2\pi T}\right)^{d/2} \exp\left(-\frac{mV^2}{2T}\right),$$

$$\left(\partial_t + \mathbf{v} \cdot \nabla\right) f(\mathbf{v}) = -\beta(\alpha)\nu_0[f(\mathbf{v}) - f_0(\mathbf{v})] + \frac{1}{2}\zeta_0(\alpha)\frac{\partial}{\partial\mathbf{v}}\cdot[\mathbf{V}f(\mathbf{v})]\right)$$

[J.J. Brey, J.W. Dufty & A.S., J. Stat. Phys. **97**, 281 (1999).] However, Prandtl number in the elastic limit: $\Pr = 1 \neq \frac{d-1}{d}$.

• Ellipsoidal statistical (ES) model:

$$\begin{aligned} \left(\partial_t + \mathbf{v} \cdot \nabla\right) f(\mathbf{v}) &= -\beta(\alpha) \frac{d-1}{d} \nu_0 [f(\mathbf{v}) - f_R(\mathbf{v})] + \frac{1}{2} \zeta_0(\alpha) \frac{\partial}{\partial \mathbf{v}} \cdot \left[\mathbf{V} f(\mathbf{v})\right], \\ f_R(\mathbf{v}) &= n \left(\frac{mn}{2\pi}\right)^{d/2} \left(\det \mathsf{R}\right)^{-1/2} \exp\left(-\frac{mn}{2} \mathsf{R}^{-1} : \mathbf{V} \mathbf{V}\right), \\ R_{ij} &= \frac{d}{d-1} p \,\delta_{ij} - \frac{1}{d-1} P_{ij}. \end{aligned}$$



FIG. 13: Time evolution of the temperature and the pressure tensor.

Mixtures

• Boltzmann equation:

$$(\partial_t + \mathbf{v}_1 \cdot \nabla) f_i(\mathbf{v}_1) = \sum_j J_{ij}^{(\alpha_{ij})}[\mathbf{v}_1|f_i, f_j],$$

$$J_{ij}^{(\alpha_{ij})}[\mathbf{v}_1|f_i, f_j] = \sigma^{d-1} \int \mathrm{d}\mathbf{v}_2 \int \mathrm{d}\widehat{\boldsymbol{\sigma}} \,\Theta(\mathbf{v}_{12} \cdot \widehat{\boldsymbol{\sigma}})(\mathbf{v}_{12} \cdot \widehat{\boldsymbol{\sigma}}) \left(\alpha_{ij}^{-2}\widehat{b}_{ij}^{-1} - 1\right) f_i(\mathbf{v}_1) f_j(\mathbf{v}_2),$$

$$\widehat{b}_{ij}\mathbf{v}_1 = \mathbf{v}_1 - \mu_{ji}(1 + \alpha_{ij})(\mathbf{v}_{12} \cdot \widehat{\boldsymbol{\sigma}})\widehat{\boldsymbol{\sigma}} \\ \widehat{b}_{ij}\mathbf{v}_2 = \mathbf{v}_2 + \mu_{ij}(1 + \alpha_{ij})(\mathbf{v}_{12} \cdot \widehat{\boldsymbol{\sigma}})\widehat{\boldsymbol{\sigma}}$$

$$\Longrightarrow$$

$$\widehat{b}_{ij}\mathbf{v}_{12} \cdot \widehat{\boldsymbol{\sigma}} = -\alpha_{ij}\mathbf{v}_{12} \cdot \widehat{\boldsymbol{\sigma}},$$

$$\mu_{ij} \equiv \frac{m_i}{m_i + m_j}.$$

• "Equivalent" system of elastic particles:

$$\underbrace{J_{ij}^{(\alpha_{ij})}[\mathbf{v}|f_i, f_j]}_{\text{inelastic collisions}} \to \beta_{ij}(\alpha_{ij}) \underbrace{J_{ij}^{(1)}[\mathbf{v}_1|f_i, f_j]}_{\text{elastic collisions}} + \underbrace{\frac{1}{2} \zeta_{ij}(\alpha_{ij}) \frac{\partial}{\partial \mathbf{v}} \cdot \left[(\mathbf{v} - \mathbf{u}_i) f_i(\mathbf{v}) \right]}_{\text{friction}}$$

• Simplest choice:

$$\beta_{ij}(\alpha_{ij}) = \frac{1 + \alpha_{ij}}{2},$$

$$\zeta_{ij}(\alpha_{ij}) = \frac{d+2}{4d} \nu_{ij}(1 - \alpha_{ij}^2),$$

$$\nu_{ij} \propto n_j \mu_{ji}^2 \sigma_{ij}^{d-1} \left(\frac{2T_i}{m_i}\right)^{1/2} \left(1 + \frac{m_i T_j}{m_j T_i}\right)^{3/2}$$

• This choice preserves the collision integrals

$$\int \mathrm{d}\mathbf{v} \,\left\{ \begin{array}{c} \mathbf{v} \\ v^2 \end{array} \right\} J_{ij}^{(\alpha_{ij})}[\mathbf{v}|f_i, f_j]$$

in the (multi-temperature) local equilibrium approximation.

• As before, kinetic models (e.g. Gross–Krook, Garzó–Santos–Brey, Andries– Aoki–Perthame, ...) for *elastic* mixtures can be extended to *inelastic* mixtures.

V. Conclusions

- A system of elastic hard spheres with a friction force succeeds in capturing the main nonequilibrium *transport* properties of a granular gas (at least in a coarse-grained way).
- To disguise as a granular gas, the elastic particles must reduce their collision rate by a factor $\beta(\alpha)$.

This is equivalent to assume that the EHS have a size smaller than the IHS: $\sigma_{\text{EHS}} = \beta^{1/(d-1)} \sigma_{\text{IHS}}.$

• It is sufficient to take

$$\beta(\alpha) = \frac{1+\alpha}{2}$$

and the *local equilibrium* approximation

$$\gamma = \frac{1}{2}\zeta_0(\alpha).$$

- Of course, the "equivalent" system of EHS is unable to retain finer details of the true system of IHS (e.g., high energy tails, velocity correlations,...).
- Further work:
 - Study of other nonequilibrium states.
 - Extension to *dense* granular gases (Enskog equation).